

## Prescribed Focus Areas

| <b>Big Idea</b>              | <b>PFA's</b>   |
|------------------------------|--|
| Searching for a better model | <ul style="list-style-type: none"><li>• Nature</li><li>• History</li></ul>   |
| The birth of Quantum Physics | <ul style="list-style-type: none"><li>• History</li><li>• Nature</li></ul>   |
| Nuclear Physics              | <ul style="list-style-type: none"><li>• Implications for society and environment</li><li>• Applications and uses</li><li>• History</li></ul>                 |
| Real world applications      | <ul style="list-style-type: none"><li>• Applications and uses</li><li>• Current issues, R&amp;D</li><li>• Implications for society and environment</li></ul> |

## Big Idea: 1. Searching for a better model

**Statement:** Problems with the Rutherford model of the atom led to the search for a model that would better explain the observed phenomena

| <b>Student activity</b>                                  | <b>Involving knowledge of</b>   |
|--|---|
| Investigate the H Spectrum (R1)                          | <ul style="list-style-type: none"><li>• Structure of the Rutherford model (L1)</li></ul>  |
| Illustrate Bohr's explanation of the Balmer series (R2)  | <ul style="list-style-type: none"><li>• Significance of H spectrum in development of Bohr's model (L2)</li><li>• Bohr's Postulates (L3)</li></ul> |
| Use Bohr's equation (R3)                                 | <ul style="list-style-type: none"><li>• Plank's quantised energy (L4)</li><li>• Development of mathematical model (L5)</li></ul>                  |
| Look at difficulties with the Rutherford-Bohr model (R4) | <ul style="list-style-type: none"><li>• Limitations of Bohr's model (L6)</li></ul>  |

## Big Idea: 2. The birth of Quantum Physics

**Statement:** The limitations of classical physics gave birth to quantum physics.

| <b>Student activity</b>  | <b>Involving knowledge of</b>  |
|--|--|
| Solve problems and analyse information using de Broglie's wave equation (R1) | <ul style="list-style-type: none"><li>• de Broglie's wave-particle duality (L1)</li><li>• Davisson and Germer's confirmation of de Broglie's proposal (L3)</li><li>• De Broglie's explanation for the stability of electron orbits in Bohr's atom (L4)</li></ul> |
| Investigate Heisenberg, Pauli contribution to atomic theory (R2)             | <ul style="list-style-type: none"><li>• Heisenberg's uncertainty principle</li><li>• Pauli's spin</li></ul>  |
| Do Electron diffraction and Taylor's experiment                              | <ul style="list-style-type: none"><li>• Interference occurs between diffracted waves (L2)</li></ul>  |

## Big Idea: 3. Nuclear Physics

**Statement:** The work of Chadwick and Fermi in producing artificial transmutations led to practical applications of nuclear physics

| Student activity  | Involving knowledge of   |
|---|--|
| Observe radiation using a cloud chamber or similar device (R1)                      | <ul style="list-style-type: none"> <li>• Components of the nucleus (L1)</li> <li>• Chadwick’s discovery of neutron, importance of conservation laws (L2)</li> </ul>  |
|   | <ul style="list-style-type: none"> <li>• Contributions of electrostatic and gravitational forces between nucleons (L7)</li> <li>• Need for the strong force (L8)</li> </ul>  |
| Do mass defect and energy calculations for transmutation and fission reactions (R2) | <ul style="list-style-type: none"> <li>• What is transmutation (L3)</li> <li>• Transmutations due to natural radioactivity (L4)</li> <li>• Need for a neutrino, Pauli’s prediction (L6)</li> </ul>                                       |
|   | <ul style="list-style-type: none"> <li>• Mass defect, Eintsein’s equivalence principle (L9)</li> </ul>   |
|   | <ul style="list-style-type: none"> <li>• Fermi’s observation of fission (L5)</li> <li>• Fermi’s demonstration of controlled chain reaction (L10)</li> <li>• Requirments for controlled and uncontrolled chain reactions (L11)</li> </ul> |

## Big Idea: 4. Real World Applications

**Statement:** An understanding of the nucleus has led to large science projects and many applications

| Student activity  | Involving knowledge of   |
|---|--|
| Assess the significance of the Manhattan project (R1)           | <ul style="list-style-type: none"><li>• Basic principles of fission reactor (L1)</li></ul>   |
| Look at isotope use in medicine, agriculture & engineering (R2) | <ul style="list-style-type: none"><li>• Medical and industrial application of isotopes (L2)</li></ul>  |
|   | <ul style="list-style-type: none"><li>• Standard model (L5)</li><li>• Particle accelerators to investigate the structure of matter (L4)</li></ul>                        |
|   | <ul style="list-style-type: none"><li>• Neutron scattering as a probe (L3)</li><li>• Paul's helium microscope</li><li>• String theory</li><li>• Nanotechnology</li></ul> |